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The vibrating molecules emitting the radiation could have only certain discrete amounts of energy, given by $E_n = nhf$, where n is called a quantum number, F is the frequency, and h is Planck's constant given by $h = 6.626 \times 10^{-34}$ joule-seconds. Molecules emit energy in units called quanta, now

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Z. V. $\int |\psi(r,t)|^2 dr = 0$. (6) This result is proven to be in agreement with the Schrödinger equation, and it then becomes natural to define the probability

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density ρ and probability current density j according to $\rho(r,t) = |\psi(r,t)|^2$, (7) $j(r,t) = \text{Re}.$

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Quantum mechanics is the branch of
physics dealing with physical

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Phenomena at microscopic scales, particularly with notions of probability. Probability in Matter. Considering light as particles (photons), the probability per volume of finding a photon in a given region of space at a given time is proportional to the number N of photons per unit volume at that time

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that it is crucial to a fundamental understanding of physics, ...

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Relativity Relative Velocity Velocity is
relative to an observer's frame of
reference. Consider 2 observers: one
in boat and one on shore watching
boat. Observer in boat would look at

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People and notice how fast he is moving relative to it. Observer on shore would look at water and boat and notice how they appear faster. Relative velocity of object can be calculated by adding up all the separate velocity vectors.

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