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hydrochloride in presence of excess of HCl which is cooled by an ice-bath such that the temperature of the reaction remains below 5 °C diazotization of primary aromatic amine occurs.  $\text{NH}_2$   $\text{NaNO}_2$   $\text{HCl}$ , < 5 °C  $\text{N}_2$   $\text{Cl}$  Mechanism

*Lecture 16 Aromatic Diazonium Salts 2 7.1.1 The Formation ...*

Lecture 16 Aromatic Diazonium Salts 711 The Formation of Diazonium Salts Addition of aqueous solution of  $\text{NaNO}_2$  to a solution of amine hydrochloride in presence of excess of HCl which is cooled by an ice-bath such that the temperature of the reaction remains below 5 °C diazotization of primary aromatic amine occurs.  $\text{NH}_2$   $\text{NaNO}_2$   $\text{HCl}$ , < 5 °C  $\text{N}_2$  ...

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Feb 16–20 Conjugated system and aromatic compounds (13, 14)) preparation and reactions of aromatic diazonium salts 8 Mar 9– 13 Aldehydes and ketones (16) Synthesis of aldehydes and ketones (25 pts each) and four lecture exams (100 pts each) will be administered during the term when assigning the final grade for the course Lecture

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### *Lecture 16 Aromatic Diazonium Salts 2 7 1 1 The Formation*

Diazonium Salts Preparation. The process of converting an organic compound, generally primary aromatic amines, into diazonium salts is referred to as diazotization or dissociation. It has been seen that the diazonium group is highly unstable under normal conditions and hence it is not generally stored, most of the time it is used just after its preparation. One of the most common methods of preparation of diazonium salt is by reacting nitrous acid with aromatic amines.

### *Diazonium Salts - Preparation, Properties & its Importance*

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### *Notes (7) - NPTEL Chemistry Principles of Organic ...*

UCI Chem 51C Organic Chemistry (Spring 2012) Lec 15. Organic Chemistry -- Hofmann Degradation, Diazotization, & Aryl Diazonium Salts -- View the complete cou...

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