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CHAPTER 17 REVIEW Reaction

Kinetics MIXED REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1. The reaction for

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17 the decomposition of hydrogen peroxide is $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$. List three ways to speed up the rate of decomposition. For each one, briefly explain why it is effective, based on collision theory.

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01:16. Problem 1 What idea is the kinetic-
molecular theory based on? ... Problem 17
Explain the relationship between
atmospheric pressure and the actual
boiling point of a liquid. Rabia B.

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CHAPTER 12 & 13: SOLUTIONS The
following pages contain the bulk (but not
all) of the information for the chapter 12 &
13 test. Focus on this content, but make
sure to review class notes, activities,
handouts, questions, etc.

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