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Correct Answer: A. Substance

A A chemist mixes 500 g of lead at 500°C with 1,200 g of water at 20°C . She then mixes 500 g of copper at 500°C with 1,200 g of water

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Lab at 20°C . The specific heat capacity of lead is $0.1276 \text{ J/g}^{\circ}\text{C}$ and the specific heat capacity of copper is $0.3845 \text{ J/g}^{\circ}\text{C}$.

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Calorimetry Lab Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature

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Lab values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after

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Correct Answer: C. The final temperature of the lead-water system will be lower than the final temperature of ...

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is an insulated container
filled with a liquid,

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