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Chapter 1 Section 2 MATTER and Its Properties Matter – Anything that has MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element. ELEMENT– a pure substance made up of only one type of atom.

MATTER and CHANGE Chapter 1 Section 1

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Properties of Matter pages 70–75 Problem-Solving Lab 1.
Explain why the flow of a compressed gas must be controlled for practical and safe use.

Matter—Properties and Changes
Matter—Properties and Changes

Chapter 1 VIDEODISC Cosmic Chemistry Greenhouse Effect, Movie Ozone Studies, Movie Nylon, Movie Dr. Jekyll and Mr. Hyde, Movie Lab Safety, Movie CD-ROM Chemistry: Matter and Change Magic of Chemistry, Demonstration The following multimedia for this chapter are available from

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Chapter 1: Introduction to Chemistry

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Law Section 6.1 Development of the Modern Periodic Table

1. octaves 2. eight 3. nine 4. accepted 5. Dmitri Mendeleev 6. atomic mass 7. elements 8. atomic number 9. Henry Moseley 10. protons 11. periodic law 12. properties 13. 14 ...

Ch 6 Study Guide answers

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Change • Chapter 5 25 Electrons in Atoms Reviewing

Vocabulary Match the definition in Column A with the term

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in Column B. Column A Column B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

Chapter Assessment

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Chemistry: Chapter 2: Matter and Change. extensive property. intensive property. extensive properties

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(examples) intensive property (examples) a property that depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not.... mass, volume, length, weight.

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Math Skills Transparency Masters Chemistry: Matter and

Change • Chapter 11 31 number of moles of known

substance C H C H C H O O O C O C O C O H O H O H O number

of moles of unknown substance O C O H O C H C O H O C H O

H O C H O C O use mole ratio 5 mol O 2 1 mol C 3 H 8 3 mol

CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol C 3 H 8 1 mol C 3 H 8 5

mol O 2

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A “ chemical change ” is a change that produces matter with a different composition than the original matter.

Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different properties than their component elements.

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