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Decomposition chemical reaction is the reaction where only one compound decomposes and results in two or more than two products. $\text{Pb}(\text{NO}_3)_2 \rightarrow \text{PbO} + 2\text{NO}_2 + \text{O}_2$. In this equation, lead nitrate is being decomposed, which breaks down to form nitrogen dioxide, oxygen, and lead oxide.

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Moles of HCl = Moles of NaOH = 0.05 L * 2 mol/L = 0.1 moles
T = 12.86 C
m (water) = 100 ml * 1.03 g/ml = 103 g
q(reaction) = -m * T * C(water) = - 103 view the full answer

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