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9/10 Part 1: Ionic vs. Covalent ~~CHEM 1411 Lecture~~

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CHEM 1411 Chapter 2 Homework Answers 1. Circle

the molecules in the following list. H₂ HCl P₄ H₂O Fe

S₈ CO₂ Ne 2. Circle the compounds in the following

list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 3. Label the

following as ionic or covalent: HCl - covalent P₄ -

covalent H₂O - covalent NaCl - ionic Ca₃(PO₄)₂ - ionic

CO₂ - covalent BaS - ionic 4. Predict the charge each

element below would prefer to have ...

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CHEM 1411. Chapter 2. Atoms, Molecules and Ions

(Homework) 1.

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CHEM 1411. Chapter 2. Atoms, Molecules and Ions (Homework)

CHEM 1411 Chapter 2 Homework Answers 1. Circle the molecules in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 2. Circle the compounds in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 3. Label the following as ionic or covalent: HCl - covalent P₄ - covalent H₂O - covalent NaCl - ionic Ca₃(PO₄)₂ - ionic Page 4/8. Download Ebook Chem 1411 Chapter 2 Homework Answers CO₂ - covalent BaS - ionic 4 ...

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CHEM 1411 Chapter 2 Homework Answers 2 CHEM 1411. Chapter 2. Atoms, Molecules and Ions (Homework) 11. The formula for sodium sulfide is A) NaS. B) K₂S. C) Na₂S. D) Na₂S. E) SeS. 12. Almost all the mass of an atom is concentrated in the nucleus. CHEM 1411. Chapter 2. Atoms, Molecules and Ions (Homework) work on the "click and go". Each problem has to be repeated 5 Page 6/8. Where To ...

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Chem 1411 Chapter 2. STUDY. PLAY. John Dalton. 1. elements composed of extremely small particles called atoms; 2. atoms of one element are different from atoms of all other elements 3. compounds are composed of atoms of more than one element; 4. chemical reaction involves only the separation, combinations, or rearrangement of atoms; not creation or destruction . law of definite proportions ...

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1.all matter is composed of atoms. 2. atoms of one element cannot be converted into atoms of another element in a chemical reaction. 3.all atoms of an element have the same number of protons and electrons which determines the chemical behavior of the element.

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CHEM 1411 Chapter 1 Homework 1. Which of the following illustrates the concept of potential energy and which illustrates kinetic energy? (a) water stored

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Homework Answers

in a water tower. (b) A rubber band stretched around a newspaper. (c) A basketball dropping through a net. (d) A car driving down the highway. (e) A flashlight battery. 2. Which of the following processes are exothermic? Endothermic? (a ...

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1 CHEM 1411. Chapter 8. Molecular Geometry and Bonding Theories (Homework) W CHEM 1411. Chapter 8. Molecular Geometry and Bonding Theories (Homework) W Multiple Choice Identify the choice that best completes the statement or answers the question. ____ 1. The electrons in the outer shell of an atom are involved in bonding. Another name for the ...

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