

Chapter 5 Test The Periodic Law

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name_____ HPS _____ # _____ date:_____ Chapter 5 TEST: The Periodic Table Multiple Choice Identify the choice that best completes the statement or answers the question. _____ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the ...

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Chapter 5 TEST: The Periodic Table Multiple Choice Identify the choice that best completes the statement or answers the question. _____ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the nucleus. d. atomic mass. _____2.

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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required

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to remove an electron from a neutral atom is the atom's

~~5 The Periodic Law~~

Holt McDougal Modern Chemistry Chapter Test Assessment Chapter Test B Teacher Notes and Answers 5 The Periodic Law TEST B 1. a 2. c 3. d 4. d 5. a 6. a 7. c 8. a 9. lanthanides 10. 2 11. fourth 12. transition elements 13. 32 14. valence electrons 15. electron affinity 16. electronegativity 17. ionization energy 18. 3 s² 3p⁴ 19. atomic radius 20 ...

~~Assessment Chapter Test B~~

Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. Mendeleev organized the chemical elements based on their a. symbols.

~~Assessment Chapter Test A~~

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~. 5 Chapter Test' The Periodic Table . 'Multiple Choice On t~ line at the left, write the letter of the answer that best completes each statement. D . 1. The properties of elements reoccur periodically when the elements are arranged in order of increasing . a. atomic mass. b. atomic number. c. atomic radius. .J) d. electron affinity. 2.

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_____ 5. Mendeleev is credited with developing the first successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. _____ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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Chapter 5 Bonding Test A Do Not Write on This Test!!! You may use a periodic table. Multiple Choice - Identify the choice that best completes the statement or answers the question.

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~~Chapter 5_ the periodic table answer key~~

The Periodic Law Section Quiz: Electron Configuration and the Periodic Table In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 25. 1. An element that has the electron configuration [Ne]3s 3p 1S in Period 25. 2.

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