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Ch. 2 Essential Questions: Must be answered on a separate sheet of paper 1) Differentiate between matter and energy. 2) Explain how each of the 4 types of energy are different. 3) Draw an atom of Carbon and label its Protons, Neutrons, and Electrons. 4) Name the 4 most abundant elements found in the Human Body. 5) Compare and contrast Ionic and Covalent Bonds.

~~Chapter 2: Basic Chemistry Name: Period~~

Figure 2—1 is a diagram of an atom. Select two different colors and use them to color the coding circles and corresponding structures on the figure. Complete this exercise by responding to the questions that follow, referring

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to the atom in this figure. Insert your answers in the answer blanks provided. O Nucleus Electrons 1. 2. 3. 5. Figure ...

~~Mr. Burghardt Ponderosa High School Shingle Springs, CA~~
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elements that form the bulk of body matter. 3. Explain how
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Chapter 2 Basic Chemistry. Objectives. Differentiate clearly

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between matter and energy. List the major energy forms, and provide one example of how each energy form is used in the body. List four...

~~Chapter 2 Basic Chemistry – Anatomy and Physiology~~
Basic Chemistry Concepts of Matter and Energy Composition of Matter Molecules and Compounds Chemical Bonds and Chemical Reactions Biochemistry. Concepts of Matter and Energy. Matter. Matter is anything that occupies space and has weight. Matter exist in three states: gas, liquid, and solid.

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1 Chapter 2.1, 2.2 Review Packet – Answer Key 2.1 The Nature of Matter Lesson Summary Atoms The atom is the basic unit of matter, made up of three subatomic particles. Protons have a positive charge and neutrons carry no charge. Strong forces bind protons and neutrons together in the nucleus. An electron is a negatively charged particle that has only about 1/1840 the mass of a proton.

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